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Intermediates in the Synthesis of *Daphni-phyllum* Alkaloids. I. A Tetracyclic Oxygen Analogue

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Abstract

In methyl (3*S*,5*S*,6*S*,10*R*,11*R*)-2-bromo-3,6-epoxy-8-iso-propyl-5-methyl-9-oxotricyclo[9.3.0.0^{5,10}]tetradeca-1,7-diene-10-carboxylate, C₂₀H₂₅BrO₄, the seven-membered ring is half-chair shaped, the six-membered ring adopts a boat conformation and the five-membered rings both adopt envelope conformations. The molecule adopts an overall hemispherical conformation.

Comment

In our projected synthesis of *Daphniphyllum* alkaloids, such as methyl homodaphniphyllate, (1), we have developed a route to an oxygen analogue, (2), containing many of the same structural features. We have previously presented the X-ray structure determination of a tetracyclic intermediate, (3) (Castellano, Brocksom & Ceschi, 1994), and this compound undergoes a com-

plex rearrangement reaction catalyzed by silver(I) salts in polar solvents (Brocksom, Pesquero & Lopes, 1990) leading to the crystalline product (2). Although the macroscopic sample is racemic, the crystal selected by hand turned out to be a single enantiomer and is the subject of the present paper. This three-dimensional structure determination is absolutely essential for successful planning of the total synthesis and cannot be accomplished by conventional spectroscopic means such as high-field NMR measurements.

A ZORTEP (Zsolnai, 1995) illustration of the formula unit showing the relative stereochemistry is presented in Fig. 1 and selected bond distances and angles are given in Table 1. Compound (2) has an ether bridge in a tetracyclic structure containing fused six-, seven- and five-membered carbocyclic skeletons, analogous to the nitrogen-bridged Daphniphyllum alkaloids. The structure is heavily influenced by the ether bridge which forces the molecule to adopt an overall hemispherical conformation. The six-membered ring is in a slightly distorted boat conformation, with C6 and C9 lying at distances of 0.516 (6) and 0.408 (5) Å, respectively, from the least-squares plane through atoms C5, C7, C8 and C10; both five-membered rings adopt envelope conformations, with C13 and C5 occupying the flap positions at distances of 0.526 (7) and -0.696 (5) Å from the least-squares plane made up of the remaining four atoms in the cyclopentane and furan rings, respectively. The molecules are joined through C-H···O interactions: C11...O1ⁱ 3.410 (7), H11...O1ⁱ 2.53 Å and C11—H11···O1ⁱ 149°; C4···O1ⁱ 3.377 (8), H4B···O1ⁱ 2.43 Å and C4—H4B···O1ⁱ 166° [symmetry code: (i) -x, $\frac{1}{2} + y$, 2 - z].

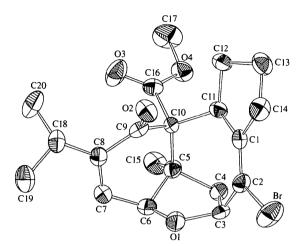


Fig. 1. The molecular structure of (2) showing the atom labelling; 50% probability displacement ellipsoids are shown.

Experimental

The synthesis of (2) has been described by Ceschi (1994). Crystals were obtained by slow evaporation from a hexane/ Et_2O mixture at 269 K.

Crystal data

$C_{20}H_{25}BrO_4$ $M_r = 409.319$ Monoclinic $P2_1$ a = 7.9210 (9) Å b = 10.6680 (10) Å c = 10.7280 (10) Å $\beta = 94.400 (10)^\circ$ $V = 903.86 (16) Å^3$ Z = 2	Mo $K\alpha$ radiation $\lambda = 0.71073 \text{ Å}$ Cell parameters from 25 reflections $\theta = 9.54-19.35^{\circ}$ $\mu = 2.296 \text{ mm}^{-1}$ T = 293 (2) K Irregular $0.38 \times 0.25 \times 0.20 \text{ mm}$ Colourless
Z = 2 $D_x = 1.5040 \text{ Mg m}^{-3}$	Colourless

Data collection

 D_m not measured

Enraf-Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: Gaussian (*PLATON*; Spek, 1990) $T_{\min} = 0.584$, $T_{\max} = 0.661$ 2402 measured reflections 2288 independent reflections

1473 reflections with $F^2 > 2\sigma(F^2)$ $R_{\text{int}} = 0.023$ $\theta_{\text{max}} = 27.97^{\circ}$ $h = -10 \rightarrow 10$ $k = 0 \rightarrow 14$ $l = 0 \rightarrow 14$ 2 standard reflections frequency: 60 min

intensity decay: 0.8%

Refinement

Refinement on F^2	$(\Delta/\sigma)_{\rm max} = 0.000$
$R[F^2 > 2\sigma(F^2)] = 0.042$	$\Delta \rho_{\text{max}} = 0.393 \text{ e Å}^{-3}$
$wR(F^2) = 0.117$	$\Delta \rho_{\min} = -0.775 \text{ e Å}^{-3}$
S = 1.077	Extinction correction: none
2288 reflections	Scattering factors from
230 parameters	International Tables for
H-atom parameters	Crystallography (Vol. C)
constrained	Absolute configuration:
$w = 1/[\sigma^2(F_o^2) + (0.0376P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$	Flack (1983)
where $P = (F_o^2 + 2F_c^2)/3$	Flack parameter = $0.00(2)$
	- ` '

Table 1. Selected geometric parameters (Å. °)

Br—C2	1.906 (5)	C1C11	1.523 (7)
O1—C6	1.439 (7)	C2—C3	1.500 (8)
O1—C3	1.457 (8)	C7C8	1.323 (8)
O2C9	1.212 (7)	C8C9	1.479 (8)
C1—C2	1.325 (8)	C8C18	1.524 (8)
C1—C14	1.499 (8)		
C6O1C3	108.5 (4)	C1—C2—Br	119.9 (4)
C2C1C14	125.9 (5)	C3C2Br	114.2 (4)
C2—C1—C11	124.9 (5)	C7C8C9	116.4 (5)
C14C1C11	109.1 (5)	C7C8C18	125.1 (6)
C1—C2—C3	125.9 (5)	C9C8C18	118.2 (5)

H atoms were refined with fixed geometry, each riding on a carrier atom, with an isotropic displacement parameter amounting to 1.5 (for methyl H atoms) or 1.2 (for the other H atoms) times the value of the equivalent isotropic displacement parameter of the atom they to which they are attached.

Data collection: CAD-4 Software (Enraf-Nonius, 1989). Cell refinement: CAD-4 Software. Data reduction: MolEN (Fair, 1990). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1985). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: ZORTEP (Zsolnai, 1995). Software used to prepare material for publication: SHELXL93.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FG1310). Services for accessing these data are described at the back of the journal.

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